

MORTGAGE RECORD.

The Herald, Lawrence, Kansas

This Indenture, Made this Second day of February, A. D. 1914, between
Sirron B. White and Maude E. White, his wife

of Douglas City County, in the State of Kansas, of the first part, and

County, in the State of Kansas, of the second part:

WITNESSETH, That said party John of the first part, in consideration of the sum of _____ and _____ DOLLARS

the receipt of which is hereby acknowledged, do _____ by these presents, Grant, Bargain, Sell, and Convey unto said part of of the second part, _____
heirs and assigns, all the following-described real estate, situated in Douglas County and State of Kansas, to wit: _____

Lot Nine (9) Block Fourteen (14) in University Place, an addition to the City of Lawrence

TO HAVE AND TO HOLD THE SAME, Together with all and singular the tenements, hereditaments and appurtenances thereunto belonging or in anywise appertaining, forever.

PROVIDED, ALWAYS, And these presents are upon this express condition, that, whereas, said
Simon W. White and Maudie E. White, his wife

has 10 this day executed and delivered one certain promissory note in writing to said part of of the second part, of which the following is a copy of :

Three years after date we promise to pay to
E. M. Bellows - Two hundred Dollars
with 6% interest from date payable semi annual
at Lawrence Kansas
Due Feb. 2, 1917

NOW, if said part CEL of the first part shall pay or cause to be paid to said part of of the second part, his heirs or assigns, said sum of money in the above-described note mentioned, together with the interest thereon, according to the terms and tenor of the same, then these presents shall be wholly discharged and void; and otherwise shall remain in full force and effect. But if said sum of sums of money, or any part thereof, or any interest thereon, is not paid when the same is due; or if the taxes and assessments of every nature which are or may be assessed and levied against said premises, or any part thereof, are not paid when the same are by law made due and payable, then the whole of said sum and sums, and interest thereon, shall and by these presents become due and payable, and said part of of the second part shall be entitled to the possession of said premises.

IN WITNESS WHEREOF, The said part III of the first part ha. th herunto set their hands^s, the day and year first above written.

State of Kansas, *Douglas* County, ss.

BE IT REMEMBERED That on this 11 day of Feb, A. D. 1944 before me, the undersigned,

17621

20 May 1964

1 Simon A. White & Maude E. White, husband

4 miles

It is a common knowledge among the people of the world that the pen is mightier than the sword, and that the instrument of writing, and not the sword, is the true power of the world.

...personally known ... who we have personally interviewed ...
...acknowledged the execution of the same.

[Handwritten signatures and stamps at the bottom of the page]

last thing written.

Notary Public

The Commission
March 5, 1966.

1881

Filed for Record on the 9th day of July, A. D. 1914 at 11 o'clock A. M.

7/1/58

14-00000

By Wm. E. T. T. T. Deputy

Figure 1. A schematic diagram of the experimental setup. The subject is seated in a chair and views the screen through a mirror. The screen displays the target (a red dot) and the starting position (a black dot). The subject's hand is positioned at the starting position. The distance between the starting position and the target is 10 cm. The subject is instructed to move the hand from the starting position to the target. The distance between the starting position and the target is 10 cm. The subject is instructed to move the hand from the starting position to the target. The distance between the starting position and the target is 10 cm.

Figure 1. The effect of the initial concentration of the monomer on the polymerization of *l*-lysine. The reaction conditions were: $[H_2O] = 100$ mol/L, $[K_2S_2O_8] = 0.005$ mol/L, $[K_2CO_3] = 0.005$ mol/L, $[K_2HPO_4] = 0.005$ mol/L, $[K_2H_2P_2O_7] = 0.005$ mol/L, $[K_2H_6P_4O_{13}] = 0.005$ mol/L, $[K_2H_8P_6O_{19}] = 0.005$ mol/L, $[K_2H_{10}P_7O_{24}] = 0.005$ mol/L, $[K_2H_{12}P_8O_{28}] = 0.005$ mol/L, $[K_2H_{14}P_9O_{32}] = 0.005$ mol/L, $[K_2H_{16}P_{10}O_{36}] = 0.005$ mol/L, $[K_2H_{18}P_{11}O_{40}] = 0.005$ mol/L, $[K_2H_{20}P_{12}O_{44}] = 0.005$ mol/L, $[K_2H_{22}P_{13}O_{48}] = 0.005$ mol/L, $[K_2H_{24}P_{14}O_{52}] = 0.005$ mol/L, $[K_2H_{26}P_{15}O_{56}] = 0.005$ mol/L, $[K_2H_{28}P_{16}O_{60}] = 0.005$ mol/L, $[K_2H_{30}P_{17}O_{64}] = 0.005$ mol/L, $[K_2H_{32}P_{18}O_{68}] = 0.005$ mol/L, $[K_2H_{34}P_{19}O_{72}] = 0.005$ mol/L, $[K_2H_{36}P_{20}O_{76}] = 0.005$ mol/L, $[K_2H_{38}P_{21}O_{80}] = 0.005$ mol/L, $[K_2H_{40}P_{22}O_{84}] = 0.005$ mol/L, $[K_2H_{42}P_{23}O_{88}] = 0.005$ mol/L, $[K_2H_{44}P_{24}O_{92}] = 0.005$ mol/L, $[K_2H_{46}P_{25}O_{96}] = 0.005$ mol/L, $[K_2H_{48}P_{26}O_{100}] = 0.005$ mol/L, $[K_2H_{50}P_{27}O_{104}] = 0.005$ mol/L, $[K_2H_{52}P_{28}O_{108}] = 0.005$ mol/L, $[K_2H_{54}P_{29}O_{112}] = 0.005$ mol/L, $[K_2H_{56}P_{30}O_{116}] = 0.005$ mol/L, $[K_2H_{58}P_{31}O_{120}] = 0.005$ mol/L, $[K_2H_{60}P_{32}O_{124}] = 0.005$ mol/L, $[K_2H_{62}P_{33}O_{128}] = 0.005$ mol/L, $[K_2H_{64}P_{34}O_{132}] = 0.005$ mol/L, $[K_2H_{66}P_{35}O_{136}] = 0.005$ mol/L, $[K_2H_{68}P_{36}O_{140}] = 0.005$ mol/L, $[K_2H_{70}P_{37}O_{144}] = 0.005$ mol/L, $[K_2H_{72}P_{38}O_{148}] = 0.005$ mol/L, $[K_2H_{74}P_{39}O_{152}] = 0.005$ mol/L, $[K_2H_{76}P_{40}O_{156}] = 0.005$ mol/L, $[K_2H_{78}P_{41}O_{160}] = 0.005$ mol/L, $[K_2H_{80}P_{42}O_{164}] = 0.005$ mol/L, $[K_2H_{82}P_{43}O_{168}] = 0.005$ mol/L, $[K_2H_{84}P_{44}O_{172}] = 0.005$ mol/L, $[K_2H_{86}P_{45}O_{176}] = 0.005$ mol/L, $[K_2H_{88}P_{46}O_{180}] = 0.005$ mol/L, $[K_2H_{90}P_{47}O_{184}] = 0.005$ mol/L, $[K_2H_{92}P_{48}O_{188}] = 0.005$ mol/L, $[K_2H_{94}P_{49}O_{192}] = 0.005$ mol/L, $[K_2H_{96}P_{50}O_{196}] = 0.005$ mol/L, $[K_2H_{98}P_{51}O_{200}] = 0.005$ mol/L, $[K_2H_{100}P_{52}O_{204}] = 0.005$ mol/L, $[K_2H_{102}P_{53}O_{208}] = 0.005$ mol/L, $[K_2H_{104}P_{54}O_{212}] = 0.005$ mol/L, $[K_2H_{106}P_{55}O_{216}] = 0.005$ mol/L, $[K_2H_{108}P_{56}O_{220}] = 0.005$ mol/L, $[K_2H_{110}P_{57}O_{224}] = 0.005$ mol/L, $[K_2H_{112}P_{58}O_{228}] = 0.005$ mol/L, $[K_2H_{114}P_{59}O_{232}] = 0.005$ mol/L, $[K_2H_{116}P_{60}O_{236}] = 0.005$ mol/L, $[K_2H_{118}P_{61}O_{240}] = 0.005$ mol/L, $[K_2H_{120}P_{62}O_{244}] = 0.005$ mol/L, $[K_2H_{122}P_{63}O_{248}] = 0.005$ mol/L, $[K_2H_{124}P_{64}O_{252}] = 0.005$ mol/L, $[K_2H_{126}P_{65}O_{256}] = 0.005$ mol/L, $[K_2H_{128}P_{66}O_{260}] = 0.005$ mol/L, $[K_2H_{130}P_{67}O_{264}] = 0.005$ mol/L, $[K_2H_{132}P_{68}O_{268}] = 0.005$ mol/L, $[K_2H_{134}P_{69}O_{272}] = 0.005$ mol/L, $[K_2H_{136}P_{70}O_{276}] = 0.005$ mol/L, $[K_2H_{138}P_{71}O_{280}] = 0.005$ mol/L, $[K_2H_{140}P_{72}O_{284}] = 0.005$ mol/L, $[K_2H_{142}P_{73}O_{288}] = 0.005$ mol/L, $[K_2H_{144}P_{74}O_{292}] = 0.005$ mol/L, $[K_2H_{146}P_{75}O_{296}] = 0.005$ mol/L, $[K_2H_{148}P_{76}O_{300}] = 0.005$ mol/L, $[K_2H_{150}P_{77}O_{304}] = 0.005$ mol/L, $[K_2H_{152}P_{78}O_{308}] = 0.005$ mol/L, $[K_2H_{154}P_{79}O_{312}] = 0.005$ mol/L, $[K_2H_{156}P_{80}O_{316}] = 0.005$ mol/L, $[K_2H_{158}P_{81}O_{320}] = 0.005$ mol/L, $[K_2H_{160}P_{82}O_{324}] = 0.005$ mol/L, $[K_2H_{162}P_{83}O_{328}] = 0.005$ mol/L, $[K_2H_{164}P_{84}O_{332}] = 0.005$ mol/L, $[K_2H_{166}P_{85}O_{336}] = 0.005$ mol/L, $[K_2H_{168}P_{86}O_{340}] = 0.005$ mol/L, $[K_2H_{170}P_{87}O_{344}] = 0.005$ mol/L, $[K_2H_{172}P_{88}O_{348}] = 0.005$ mol/L, $[K_2H_{174}P_{89}O_{352}] = 0.005$ mol/L, $[K_2H_{176}P_{90}O_{356}] = 0.005$ mol/L, $[K_2H_{178}P_{91}O_{360}] = 0.005$ mol/L, $[K_2H_{180}P_{92}O_{364}] = 0.005$ mol/L, $[K_2H_{182}P_{93}O_{368}] = 0.005$ mol/L, $[K_2H_{184}P_{94}O_{372}] = 0.005$ mol/L, $[K_2H_{186}P_{95}O_{376}] = 0.005$ mol/L, $[K_2H_{188}P_{96}O_{380}] = 0.005$ mol/L, $[K_2H_{190}P_{97}O_{384}] = 0.005$ mol/L, $[K_2H_{192}P_{98}O_{388}] = 0.005$ mol/L, $[K_2H_{194}P_{99}O_{392}] = 0.005$ mol/L, $[K_2H_{196}P_{100}O_{396}] = 0.005$ mol/L, $[K_2H_{198}P_{101}O_{400}] = 0.005$ mol/L, $[K_2H_{200}P_{102}O_{404}] = 0.005$ mol/L, $[K_2H_{202}P_{103}O_{408}] = 0.005$ mol/L, $[K_2H_{204}P_{104}O_{412}] = 0.005$ mol/L, $[K_2H_{206}P_{105}O_{416}] = 0.005$ mol/L, $[K_2H_{208}P_{106}O_{420}] = 0.005$ mol/L, $[K_2H_{210}P_{107}O_{424}] = 0.005$ mol/L, $[K_2H_{212}P_{108}O_{428}] = 0.005$ mol/L, $[K_2H_{214}P_{109}O_{432}] = 0.005$ mol/L, $[K_2H_{216}P_{110}O_{436}] = 0.005$ mol/L, $[K_2H_{218}P_{111}O_{440}] = 0.005$ mol/L, $[K_2H_{220}P_{112}O_{444}] = 0.005$ mol/L, $[K_2H_{222}P_{113}O_{448}] = 0.005$ mol/L, $[K_2H_{224}P_{114}O_{452}] = 0.005$ mol/L, $[K_2H_{226}P_{115}O_{456}] = 0.005$ mol/L, $[K_2H_{2$

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Figure 1. The effect of the number of iterations on the accuracy of the proposed algorithm. The accuracy of the proposed algorithm increases with the number of iterations. The accuracy of the proposed algorithm is 100% when the number of iterations is 1000.

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(THE FOLLOWING IS THE FIRST TWO COLUMNS OF THE DOCUMENT)

\$500.00 Robertson June 19 1968
Received of Emil Schmidt _____ Dollars, in full
Gift from Emmer

Recorded Jan. 21st - 1916
 Noel L. Lawrence